Simulated Acidic Weathering of "Black Granites": an Assessment using Regression Analysis

Simulação de Atmosfera Ácida em "Granitos Negros": uma Avaliação por Análise de Regressão

Ely Brasil de Arruda Luna Cavalcanti¹ , José de Araújo Nogueira Neto² , Eldemar de Albuquerque Menor³ , Adejardo Francisco da Silva Filho¹ , Marcelo Reis Rodrigues da Silva¹ & Lucas Fontenele Amorim⁴

¹Universidade Federal de Pernambuco, Departamento de Geologia, Recife, PE, Brasil

²Universidade Federal de Goiás, Faculdade de Ciências e Tecnologia, Departamento de Geologia, Campus do Aparecida de Goiânia, Goiânia, GO, Brasil

³Universidade Federal do Ceará, Programa de pós-Graduação em Desenvolvimento e Meio Ambiente, Campus do Pici, Fortaleza, CE, Brasil

⁴Instituto Federal do Ceará, Camocim, CE, Brasil

E-mails: elybgeo@gmail.com; nogueiraneto.josedearaujo@gmail.com; eamenor@gmail.com; afsf@ufpe.br; marcelor@ufpe.br; lucas.amorim@ifce.edu.br

Corresponding author: José de Araújo Nogueira Neto; nogueiraneto.josedearaujo@gmail.com

Abstract

A number of "black granites" were tested using simulated acidic weathering $(H_2SO_4, HNO_3, H_2SO_4/HNO_3$ and HCl, at pH = 3,00), freeze resistance and relative loss of brightness. The same plates of these granites remained immersed in each acidic solution over eight weekly cycles. The pH data from residual acid solutions were plotted versus timing of the experimental cycles in XY diagrams for interpretation as log-linear regressions. The global set of data, transcribed in a ternary diagram, shows a distinctive area highlighting the best commercial granites. In addition to these granite coatings, the proposed methodology allows easy and accurate identification of their best performances.

Keywords: Geochemistry; Alterability; Ornamental rock

Resumo

Alguns "granitos negros" foram testados por meio de meteorização ácida simulada $(H_2SO_4, HNO_3, H_2SO_4 / HNO_3 e HCl, a pH = 3,00)$, resistência ao congelamento e perda relativa de brilho. As mesmas placas destes granitos permaneceram imersas em cada solução ácida durante oito ciclos semanais. Os dados de pH de soluções ácidas residuais foram traçados em função do tempo dos ciclos de experimentação em diagramas XY para interpretação como regressões log-lineares. O conjunto global de dados, transcritos em diagrama ternário, mostra uma área distinta destacando os melhores granitos comerciais. Fora destes revestimentos de granitos, a metodologia proposta permite uma identificação fácil e precisa dos seus melhores desempenhos

Palavras-chave: Geoquímica; Alterabilidade; Rocha ornamental

1 Introduction

The so-called "black granites" include several lithologies that have been widely commercialized for many centuries. The most common are gabbros, diabases,

basalts, anorthosites, diorites, and metamorphic rocks. *Nero* assoluto is a lithic product whose homogeneous black color is due to a predominance of mafic minerals and very fine texture. A very extensive range of black granites, contains different mineralogical compositions and textures, which

Received: 19 November 2020; Accepted: 13 April 2021 Anu. Inst. Geociênc., 2023;46:39560





can influence the performance of external cladding products, leading to unexpected and undesirable results.

In order to predict the behavior of lytic coating plates, particularly those exposed to natural weathering, several authors have investigated the individual behavior of major minerals contained in ornamental rocks (Shen et al. 2019; White & Brantley 2003). Other authors have applied accelerated weathering to rock plates in the laboratory using various acidic solutions (Simão, Leal & Galhano 2020). These processes face an as yet unsurmounted problem: the transposition of each experimental cycle in the laboratory to its corresponding real-time in nature. The difficulty lies in the limited number of variables that can be controlled together in the laboratory, as opposed to a greater number that interfere in nature, producing rock weathering under considerably different environments. More importantly, geological time is not available in the laboratory to produce results.

From a marketing standpoint, in order to overcome the difficulty of selecting products for external lytic coatings, it is assumed that relatively homogeneous black granites behave predictably and similarly against natural weathering. However, from a scientific perspective, it is known that these products may exhibit sufficiently different magmatic evolution to produce markedly different performance under meteoric conditions. This paper studies and compares the performance of some traditional Brazilian black granites marketed as being of "equal quality". This type of experimental study can also contribute to safe and correct application of the rock, avoiding esthetic, structural and negative economic impacts.

2 Samples and Methods

Accelerated acidic weathering of four Brazilian "black granites" (BG1, BG2, BG3 and BG4) was performed using fresh cut plates. These simulated weathering experiments were carried out using standard acidic solutions at pH = 3,00: H_2SO_4 , HNO_3 , H_2SO_4/HNO_3 (50/50%) and HCl, reacting on two polished plates (5 x 5 x 2 cm) of the "granite" variety. The experiments involved eight weekly cycles with samples immersed in acidic reagent

(Ribeiro 2011). A Digimed pHmeter measured the pH of each residual solution. Plates were removed from the acid solutions once a week, washed in deionized water and dried at room temperature. They were then re-submerged in a new acid solution, the same as in the previous immersion. At the beginning (t = 0), middle (t = 28 days) and end (t = 56 days), of the experiments, brightness was measured on the polished surface of the dry plates (average of 27 measurements), using a Sanwa Kenma IG-330-Gloss Checker, expressed as percentage reflectance.

The pH of the residual acidic solutions (variable y) and the cumulative time of the reactive steps (variable X) were plotted as linear regressions in order to define the "critical time" throughout the reactive experiments. The "critical time" is reached when the standard acidic reagent displays apparent ineffectiveness over a certain time step, i.e. when the pH of the residual acidic solution is the same as the acidic reagent selected. This condition indicates that hydrolytic corrosion dissolved the most weatherable surface minerals, producing a discernible loss of brightness on the polished plate. This "critical time" is graphically determined in XY diagrams, and indicates possible "granite" performance when submitted to natural weathering. The best products are those whose linear regressions reach the X axis later when Y = 3.

Petrographic analyses were performed to determine the mineralogical composition and classification of the "black granites" studied and select only fresh samples, where textures are considered negligible for an evaluation of commercialized "black granites".

3 Results and Discussion

3.1 Petrographic Data

The modal composition indicates that the selected "granites" are basic rocks (Table 1) from presumably well-preserved quarries. Thus, the influence of textural characteristics (e.g. Navarro 2002; Passchier & Trouw 2005), is not considered in this study.

| Black granite | Petrographic classification | Plagioclases | Pyroxenes | Biotite | Hornblende | Accessories (opaques) | Accessories (others) |
|------------------|-----------------------------|--------------|-----------|---------|------------|--------------------------|-------------------------|
| BG1 | Gabbro-norite | 63,1 | 8,9 | 17,3 | 6,3 | 4,4 | tr |
| BG2 | Biotite diorite | 59,8 | 14,3 | 20,4 | 3,9 | 1,6 | tr |
| BG3 | Diorite-norite | 41,8 | 24,0 | 17,7 | 8,0 | 7,2 | 1,3 |
| BG4 | Ortho-Amphibolite | 44,5 | | 2,9 | 48,4 | 3,5 | 1,7 |

Table 1 Summary on the modal composition of the studied "black granites".

Among these rocks, pyroxene participation in whole rock (WR) is always greater than that of amphibole (hornblende), while biotite is relatively constant (17.3 to 20.4% WR), except in the BG4 "granite". The most striking difference involves the number of feldspars (plagioclases), more important in the BG1 and BG2 varieties than in BG3 (41.8% WR). By contrast, the BG4 variety includes hornblende and plagioclase as dominant minerals, while pyroxene is absent and biotite an accessory mineral.

3.2 Relative Loss of Brightness

The relative loss of brightness (Table 2) reveals discrepant results because the tested plates did not necessarily originate in the same commercialized block.

This reinforces the hypothesis that even so-called homogeneous mafic rocks usually show different facies across their outcrops (e.g. Hyndman 1985; McBirney 2007). This suggests that the technical characterization

| Table 2 Summary of | of medium surface bri | ahtness of "black | aranites" (2 | 7 measurements/pla | ate) alono | a different times of th | ne weathering cycles. |
|---------------------------------------|-----------------------|-------------------|--------------|--------------------|------------|-------------------------|---------------------------------------|
| · · · · · · · · · · · · · · · · · · · | | | J (| | | | · · · · · · · · · · · · · · · · · · · |

| Black Granite | Ex | ∑ Time of Weathering perimentation / Bright | Acidic solution | % Final Relative | | |
|---------------|-------|--|-----------------|--|--------------------|--|
| | T = 0 | T = 28 days | T = 56 days | — рн = 3,00 | Loss of Brightness | |
| | 75,92 | 51,92 | 46,48 | 11 60 | 39 | |
| | 72,22 | 65,48 | 56,14 | п ₂ 30 ₄ | 22 | |
| | 71,55 | 56,81 | 53,81 | | 25 | |
| BG1 | 78,29 | 59,00 | 47,51 | | 39 | |
| | 81,88 | 59,37 | 57,59 | | 30 | |
| | 82,92 | 69,59 | 63,59 | H ₂ SO ₄ /HNO ₃ | 23 | |
| | 71,18 | 61,55 | 49,48 | | 30 | |
| | 77,03 | 60,70 | 44,70 | пы | 42 | |
| | 73,14 | 53,29 | 46,22 | 11.00 | 27 | |
| | 42,74 | 24,03 | 23,40 | H ₂ SU ₄ | 45 | |
| | 79,92 | 67,14 | 57,11 | | 29 | |
| BG2 | 83,24 | 76,55 | 66,92 | HNO ₃ | 20 | |
| | 78,85 | 71,66 | 55,03 | | 30 | |
| | 87,59 | 62,70 | 55,59 | H_2SO_4/HNO_3 | 37 | |
| | 80,66 | 58,41 | 46,29 | | 43 | |
| | 79,62 | 58,88 | 50,55 | HCI | 37 | |
| | 64,33 | 57,66 | 52,59 | | 18 | |
| | 64,74 | 61,92 | 50,18 | H ₂ SU ₄ | 22 | |
| | 48,37 | 44,03 | 42,37 | | 12 | |
| BG3 | 50,70 | 43,96 | 41,29 | HNO ₃ | 19 | |
| | 54,92 | 48,70 | 46,70 | | 11 | |
| | 64,11 | 54,37 | 50,55 | H_2SO_4/HNO_3 | 21 | |
| | 53,11 | 47,25 | 46,03 | | 13 | |
| | 79,62 | 46,77 | 46,03 | псі | 42 | |
| | 63,70 | 48,81 | 48,07 | | 25 | |
| | 64,70 | 54,48 | 48,18 | H ₂ SU ₄ | 26 | |
| | 70,14 | 55,44 | 51,18 | | 27 | |
| BG4 | 45,40 | 34,22 | 31,11 | HNO ₃ | 31 | |
| | 74,33 | 62,59 | 55,18 | | 26 | |
| | 58,74 | 41,14 | 38,62 | H2504/HNO3 | 34 | |
| | 52,55 | 39,77 | 35,25 | | 33 | |
| | 62,96 | 43,74 | 38,81 | HUI | 38 | |

of ornamental rocks is significantly dependent on the representativeness of their tested samples, and often links their characterization to particular facies of a lithic body.

Considering the overall relative loss of brightness (n = 8/tested variety), whose data come from different acid weathering simulations, the performance of most "black granites" would be similar, reaching around 30% loss over their original brightness after 56 days of experimentation (Table 3). By contrast, the "BG3 variety" would be a noteworthy exception and its lower vulnerability to weathering suggests better quality in the tested rocks. This statement applies to mafic rock with greater pyroxene WR participation, i.e., with a corresponding lower proportion (67.5%) of aluminosilicate minerals. Moreover, overall relative loss of brightness shows different responses from different rocks commercialized as "black granite". The stability of minerals against weathering differs between pyroxenes, amphiboles, and biotites, following old concepts established in Bowen's reaction series, later ratified by Goldich (1938). However, regardless of the mineral association, when crystalline structures are physically affected, acidic reactivity will be more effective and faster in these cases. This produces micrometric scraping on the plate surface, causing dispersion in the reflection of incident light that results in a faster decline in brightness.

In agreement, the results showed that the relative loss of brightness is more intense during the first weathering steps (28 days), tending almost invariably to subsequent asymptotic behavior (Figures 1A, B, C and D). This behavior is generally adjustable as first-order linear regression. However, original well-polished plates (higher brightness values) show relative loss of brightness as a logarithmic regression design.

3.3 Simulated Acidic Weathering

Several studies on simulated acidic weathering and the weathering of silicate minerals in nature have been conducted for several decades.

Since silicate minerals are the most abundant in the Earth's crust, their reactive kinetics have strong repercussions on the hydrochemistry of surface waters, as shown by Gibbs (1970), Meybeck (1987), Velbel (1993), Viers et al. (1997) and White and Brantley (2003), among many others. Several factors have influenced the reactive kinetics of major rock-forming silicates (Ollier 1984), mainly their crystallographic nature (Velbel 1999; Brantley 2008), regional rainfall hydrochemistry, climate (White & Blum 1995; Viers et al. 1997), and even the geological time scale (White & Brantley 2003).

Research involving acid attacks on gabbroic natural stones (commercialized as "black granite" because of their lytic nature) can be found in Simão and Silva (1997), Silva and Simão (1997, 2003) and Simão and Carvalho (2005), where the rocks were attacked by acidic solutions in order to observe their alterations as a response to weathering in polluted environments. These changes in mineralogical composition were reported by Silva and Simão (2004), in chemical composition by Galembeck et al. (2009) and Simão and Silva (2003) and in chromatic changes by Galembeck et al. (2009).

In simulated acidic weathering, only a reduced set of variables are considered and jointly controlled. It is deemed impossible to reproduce all possible variables acting in nature in the laboratory, because these variables work over a geological time scale. Nevertheless, the results tend to corroborate the mineral vulnerability predicted by

| "Black Granite" | Petrographic | Some mineral | Original brightness | Final relative loss of brightness |
|-----------------|-------------------|------------------------------|------------------------|--------------------------------------|
| | classification | components (%) | AA SD | AA SD |
| BG1 | Gabbro-norite | Pyrox. = 8,9 Anf. = 6,3 | 76,4 4,6 | 31,3 7,9 |
| BG2 | Biotite diorite | Pyrox. = 14,3 Anf. = 3,9 | 75,7 13,9 | 33,5 8,5 |
| BG3 | Diorite-norite | Pyrox. = 24,0 Anf. = 8,0 | 60,0 10,3 | 19,8 9,9 |
| BG4 | Ortho-amphibolite | Pyrox. = none Anf. = 48,4 | 61,6 9,3 | 30,0 4,7 |

Table 3 Statistical data about the original brightness and relative loss of brightness of some Brazilian "black granites", under simulated acidic weathering (56 days).

AA = arithmetic average

SD = standard deviation

Goldich (1938). The best performance of "granite" BG-3, with greater participation of pyroxenes WR, may be due to silica precipitation from these minerals, as demonstrated by Schott et al. (1981) at low pH solutions. This mechanism would produce a protective film on the polished plates, delaying its loss of brightness.

The pH values from the simulated acidic weathering steps (Table 4) are presented in multiple scatterplot diagrams, showing their average values (two tested polished plates through accelerated alterability) versus the cumulative time in the corresponding experiment (Figures 2A, B, C and D).

A first approach to the pH values in scatterplot diagrams confirms that their best adaptable configuration is by log-linear regression. The evidence of higher initial reactivity provoked by the acidic solutions follows that observed during the experiments on relative loss of brightness, that is, regardless of the mineral composition of the rocks, procedures for cutting and polishing the plates may cause deformations in crystallographic structures, generating major reactive surface vulnerability to hydrolysis. The tendency to asymptotic behavior during the subsequent reactive cycles represents the time when an affected crystalline surface is being removed by acid etching.

During the first cycles of reactive kinetics, the number of minerals showing major vulnerability to weathering diminishes on the plate surface. Thus, the consumption of the reagent acid solution declines, considering the short duration of each cycle. The loglinear regression can reach the X axis for pH = 3,00 (Y variable), featuring the same pH values between standard acid solutions and their corresponding residual acidic solutions. From a methodological standpoint, this represents a "critical time" in the weathering experiment, graphically determinable or calculated using the statistical equation of the log-linear regression. The earlier the critical time is reached, the more vulnerable the black granite tested. Pragmatically, at this time point in the simulated experiment, the weathering consequences on the polished plate are well known, affecting its luster and creating an undesirable esthetic aspect, considered unacceptable. As such, this statistical treatment provides values to assess and compare the vulnerability of ornamental rocks under meteoric weathering.



Figure 1 Relative loss of brightness measured in polished plates of granites: A. BG1; B. BG2; C. BG3; D. BG4; during simulated weathering experiments, using different acidic solutions at pH = 3.00.

| | Table 4 Summary | of pH values from | different residua | l acidic solutions alon | g simulated eathering | g of some Black g | granites. |
|--|-----------------|-------------------|-------------------|-------------------------|-----------------------|-------------------|-----------|
|--|-----------------|-------------------|-------------------|-------------------------|-----------------------|-------------------|-----------|

| | BG1 | | | | | | | | | | | В | G2 | | | |
|----------|------------------|--------------|-----|-----|--------------------------------|-------------------|-----|-----|--------------------------------|-----|------------------|-----|--|-----|-----|-----|
| Maska | H ₂ S | 5 0 4 | н | 10, | H ₂ SO ₄ | /HNO ₃ | Н | CI | H ₂ SO ₄ | | HNO ₃ | | H ₂ SO ₄ /HNO ₃ | | HCI | |
| vveeks - | P1 | P2 | P3 | P4 | P5 | P6 | P7 | P8 | P1 | P2 | P3 | P4 | P5 | P6 | P7 | P8 |
| 1 | 4,8 | 4,6 | 4,3 | 4,5 | 4,5 | 4,0 | 4,6 | 4,2 | 4,2 | 4,2 | 4,3 | 4,2 | 4,0 | 4,0 | 4,2 | 4,3 |
| 2 | 4,4 | 4,2 | 4,0 | 4,0 | 4,1 | 3,7 | 4,2 | 3,8 | 3,8 | 3,8 | 3,8 | 3,8 | 3,6 | 3,6 | 3,7 | 3,4 |
| 3 | 4,2 | 4,0 | 3,6 | 3,7 | 4,1 | 3,8 | 4,1 | 3,9 | 3,6 | 3,7 | 3,6 | 3,6 | 3,7 | 3,7 | 3,8 | 3,8 |
| 4 | 4,0 | 3,8 | 3,6 | 3,6 | 3,8 | 3,6 | 3,9 | 3,7 | 3,7 | 3,6 | 3,6 | 3,5 | 3,5 | 3,5 | 3,6 | 3,7 |
| 5 | 3,9 | 3,7 | 3,8 | 3,8 | 3,8 | 3,8 | 3,9 | 3,7 | 3,5 | 3,5 | 3,6 | 3,5 | 3,6 | 3,5 | 3,6 | 3,6 |
| 6 | 3,7 | 3,6 | 3,4 | 3,4 | 3,6 | 3,4 | 3,6 | 3,4 | 3,4 | 3,5 | 3,3 | 3,4 | 3,3 | 3,4 | 3,4 | 3,5 |
| 7 | 3,6 | 3,5 | 3,3 | 3,3 | 3,5 | 3,4 | 3,6 | 3,5 | 3,4 | 3,4 | 3,3 | 3,3 | 3,3 | 3,3 | 3,4 | 3,4 |
| 8 | 3,6 | 3,5 | 3,4 | 3,4 | 3,4 | 3,4 | 3,6 | 3,5 | 3,4 | 3,4 | 3,3 | 3,3 | 3,3 | 3,3 | 3,3 | 3,3 |
| | | | | В | G3 | | | | | | | В | G4 | | | |

| Weeke | H ₂ S | SO ₄ | н | 10, | H ₂ SO ₄ | /HNO ₃ | н | НСІ | | HCI | | HCI | | HCI | | HCI | | HCI | | HCI | | H_2SO_4 | | HNO ₃ | | H ₂ SO ₄ /HNO ₃ | | HCI | |
|-------|------------------|------------------------|-----|-----|--------------------------------|-------------------|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|--|-----|--|-----|--|-----------|--|------------------|--|--|--|-----|--|
| Weeks | P1 | P2 | P3 | P4 | P5 | P6 | P7 | P8 | P1 | P2 | P3 | P4 | P5 | P6 | P7 | P8 | | | | | | | | | | | | | |
| 1 | 4,2 | 4,1 | 4,1 | 3,8 | 3,8 | 3,7 | 4,0 | 4,1 | 3,7 | 3,6 | 3,5 | 3,6 | 3,5 | 3,5 | 3,6 | 3,5 | | | | | | | | | | | | | |
| 2 | 3,8 | 3,7 | 3,7 | 3,6 | 3,6 | 3,6 | 3,6 | 3,7 | 3,5 | 3,5 | 3,4 | 3,4 | 3,2 | 3,2 | 3,3 | 3,4 | | | | | | | | | | | | | |
| 3 | 3,6 | 3,6 | 3,5 | 3,6 | 3,6 | 3,6 | 3,6 | 3,7 | 3,3 | 3,3 | 3,3 | 3,2 | 3,4 | 3,4 | 3,4 | 3,4 | | | | | | | | | | | | | |
| 4 | 3,4 | 3,5 | 3,5 | 3,5 | 3,5 | 3,5 | 3,5 | 3,5 | 3,2 | 3,2 | 3,2 | 3,2 | 3,2 | 3,2 | 3,3 | 3,3 | | | | | | | | | | | | | |
| 5 | 3,6 | 3,4 | 3,5 | 3,5 | 3,5 | 3,5 | 3,5 | 3,6 | 3,2 | 3,2 | 3,2 | 3,3 | 3,2 | 3,2 | 3,3 | 3,3 | | | | | | | | | | | | | |
| 6 | 3,4 | 3,4 | 3,4 | 3,3 | 3,3 | 3,3 | 3,4 | 3,5 | 3,2 | 3,2 | 3,1 | 3,1 | 3,2 | 3,2 | 3,2 | 3,2 | | | | | | | | | | | | | |
| 7 | 3,4 | 3,4 | 3,3 | 3,3 | 3,3 | 3,3 | 3,4 | 3,3 | 3,1 | 3,1 | 3,1 | 3,1 | 3,1 | 3,1 | 3,2 | 3,2 | | | | | | | | | | | | | |
| 8 | 3,4 | 3,3 | 3,3 | 3,3 | 3,3 | 3,3 | 3,3 | 3,3 | 3,2 | 3,1 | 3,1 | 3,1 | 3,1 | 3,1 | 3,1 | 3,1 | | | | | | | | | | | | | |



Figure 2 pH variations of residual acidic solutions from simulated weathering of black granites: A. BG1; B. BG2; C. BG3; D. BG4; using differents acidic solutions.

The general behavior of the rocks studied under simulated weathering is similar despite different acidic solution performance and mineralogical composition. In this context, two results can be expected: the linear regression reaches the X axis for pH = 3,00 (variable Y) before the eighth week, or it would need a longer experiment for this to occur, followed by mathematical calculations to determine the "critical time". Pragmatically, these timing values indicate rock performance when exposed to natural conditions. Accordingly, "BG3 granite" performed best, while "BG1 and BG4 granites" exhibited impairments. These conclusions corroborate freeze resistance tests (*in* Cavalcanti 2013) and the relative loss of brightness experiments (Table 5).

Given the widespread availability of "black granites" in a globalized market, practical graphical visualization can be designed through a ternary diagram containing the variables under study here (Figure 3). This diagram displays the domains that characterize the predictability of the best performances of these granites in terms of their exposure in natural settings.

 Table 5 Summary of different experiments on the qualitative evaluation of "black granites" under simulated weathering during steps totaling 56 days.

| "Black granite" | Final relative loss of brightness (%) | Timing to the critical time (weeks) (*) | Freeze resistance (% medium relative loss of weight) | | |
|-----------------|---------------------------------------|---|--|--|--|
| BG1 | 31,3 | 10,5 | 0,31 | | |
| BG2 | 33,5 | 20,0 (**) | 0,22 | | |
| BG3 | 19,8 | 17,5 (**) | 0,11 | | |
| BG4 | 30,0 | 11,0 | 0,35 | | |
| Reference | This work | This work | Cavalcanti, 2013 | | |

(*) **chosen** acidic solution: H₂SO₄ pH = 3,00

(**) calculated from a log-linear equation mathematics



Figure 3 Diagram for predictability of the behavior of "black granites" under meteoric weathering.

4 Conclusion

"Black granites" exhibit a wide lithological variation in peridotites, basalts, gabbros, gabbro-norites, and diorites, since the visual pattern is dominated by dark tones. These rocks display remarkable differences in their mineralogical composition, and different performances under acidic weathering. However, the relative behavior of their polished plates can be evaluated by comparing their performance under simulated weathering, freezeresistance experiments and loss of brightness, using standard laboratory procedures. The association of this set of parameters was selected because they produce visible damage to the polished plates.

The data produced in the present study were plotted into a triangular diagram that defines specific domains, indicating with better reliability a predictive assessment for "black granites" when exposed to acidic weathering. In terms of esthetic conservation, this methodology allows easy comparison between "black granites", leading to a more reliable assessment of these ornamental rocks. In other words, the method proposed here enables the rejection of low-performance products when selecting "black granites" available on the market, particularly where the application is for external cladding. Considering comparison criteria proposed here for "black granites", the remaining challenge is to estimate as accurately as possible the time it takes for the surface of "black granite" to exhibit unacceptable loss of esthetic quality.

5 Acknowledgments

The authors are grateful for the support of the National Council for Technological and Scientific Development (CNPq) and the Cearense Foundation for the Support of Technological and Scientific Development (FUNCAP) [process number 0024- 01082.02.00 / 25].

6 References

- Brantley, S.L. 2008, 'Kinetics of mineral dissolution', in S.L. Brantley, A.F. White & J.D. Kubicki (eds), *Kinetics of waterrock interaction*, Springer Verlag, New York, pp. 151-210, DOI:10.1007/978-0-387-73563-4_5
- Cavalcanti, E.B.A. 2013, 'Estudo comparativo de algumas rochas ornamentais máficas tipo "granito preto", por alterabilidade acelerada e gelo-degelo: uma nova abordagem geoestatística', PhD thesis, Programa de Pós-Graduação em Geografia, Universidade Federal de Pernambuco, Recife.
- Galembeck, T., Simão, J., Nogueira Neto, J.A., Artur, A.C. & Silva, Z. 2009, 'Comportamento cromático de rochas ornamentais sob envelhecimento acelerado', XXIII Simpósio Geologia do Nordeste, Sociedade Brasileira de Geologia, Fortaleza, Ceará, viewed 02 november 2020, <https://www.researchgate. net/publication/338923791_Comportamento_cromatico_de_ rochas_ornamentais_sob_envelhecimento_acelerado/citations>.
- Gibbs, R.J. 1970, 'Mechanisms controlling world water chemistry', *Science*, vol. 170, no. 3962, pp. 1088-90, DOI:10.1126/ science.170.3962.1088
- Goldich, S.S. 1938, 'A study on rock weathering', *Journal of Geology*, vol. 46, no. 01, pp. 17-58, DOI:10.1086/624619
- Hyndman, D.W. 1985, *Petrology of igneous and metamorphic rock*, McGraw Hill, New York.
- McBirney, A. 2007, *Igneous petrology*, 3nd edn, Jones & Bartlett, Ontario.
- Meybeck, M. 1987, 'Global chemical weathering of surficial rocks estimated from river dissolved loads', *American Journal of Science*, vol. 287, no. 5, pp. 401-28, DOI:10.2475/ ajs.287.5.401
- Navarro, F.C. 2002, 'Caracterização petrográfica como técnica para a previsão do comportamento físico e mecânico de "granitos" ornamentais', PhD dissertation, Instituto de Geociências e Ciências Exatas, Universidade Estadual Paulista, Rio Claro, São Paulo.
- Ollier, C. W. 1984, Weathering, 2nd edn, Longman, New York,.
- Passchier, K. & Trouw, R.A.J. 2005, *Micro-tectonics*, 2nd edn, Springer Verlag, Berlin.
- Ribeiro, W.J.M. 2011, 'Estudo comparativo de alterabilidade acelerada de três fácies comerciais do sienito Marrom Imperial', PhD dissertation, Programa de Pós-Graduação em Geografia, Universidade Federal de Pernambuco, Recife.
- Schott, J., Berner, R.A. & Sjoberg, E.L. 1981, 'Mechanism of pyroxene and amphiboleweathering-I: experimental studies of iron-free minerals', *Geochimica et Cosmochimica Acta*, vol. 45, no. 11, pp. 2123-35, DOI:10.1016/0016-7037(81)90065-X
- Silva, Z. & Simão, J. 1997, 'Acção de poluentes ácidos na superfície de rochas silicatadas: observações em anortositos ao microscópio electrónico de varrimento', X Semana de Geoquímica/IV Congresso de Geoquímica dos Países de Língua Portuguesa, Sociedade Brasileira de Geoquímica, Braga, pp. 523-24.

- Silva, Z. & Simão, J. 2003, 'Estudo da alteração de rochas ígneas ornamentais para prevenção de utilização: diagnóstico em rochas gabróicas', *3th ENCORE*, Lisboa, pp. 511-18, viewed 02 november 2020, <https://www.researchgate. net/profile/J-Simao/publication/338924140_Estudo_da_ Alteracao_de_Rochas_Igneas_Ornamentais_para_Prevencao_ de_Utilizacao_Diagnostico_em_Rochas_Gabroicas/ links/5e371c78458515072d7a06af/Estudo-da-Alteracao-de-Rochas-Igneas-Ornamentais-para-Prevencaode_Otilizacao-Diagnostico_em_Rochas_Gabroicas/ links/5e371c78458515072d7a06af/Estudo-da-Alteracao-de-Rochas-Igneas-Ornamentais-para-Prevencao-de-Utilizacao-Diagnostico-em-Rochas-Gabroicas.pdf>.
- Silva, Z. & Simão, J. 2004, 'Evaluation of alteration features on rock forming minerals – SEM observations on dimension stones', 6th International Symposium on the Conservation of Monuments in the Mediterranean Basin, Lisboa, pp. 347-51, viewed 02 november 2020, <https://www.researchgate.net/ publication/338922429_Evaluation_of_alteration_features_ on_rock_forming_minerals_-_SEM_observations_on_ dimension_stones>.
- Simão, J. & Carvalho, C. 2005, 'Ornamental stone decay in cold and polluted environments – degradation evaluation based on resonance frequency measurements and flexural strength tests', *I International Congress on Dimension Stone (CETEM/CETEMAG)*, Guarapari, Espírito Santo, pp. 01-05, viewed 02 november 2020, <https://www.researchgate. net/profile/J-Simao/publication/338922442_Ornamental_ stone_decay_in_cold_and_polluted_environments_degradation_evaluation_based_on_resonance_ frequency_measurements_and_flexural_strength_tests/ links/5e371699a6fdccd96581aa66/Ornamental-stone-decayin-cold-and-polluted-environments-degradation-evaluationbased-on-resonance-frequency-measurements-and-flexuralstrength-tests.pdf>.
- Simão, J. & Silva, Z. 1997, 'Anorthosite and its use as dimension stone: alterations as response to weathering in polluted environment', *International Symposium Engineering Geology And the Environment*, vol. 03, pp. 3257-62, viewed 02 november 2020, https://www.researchgate. net/publication/291062832_Anorthosite_and_its_use_as_ dimension_stone_alterations_as_response_to_weathering_ in_polluted_environment/figures?lo=1>.
- Simão, J. & Silva, Z. 2003, 'Mobilidade química em processos de alteração: ensaios em rochas ornamentais', *IV Congresso Ibérico Geoquímica/XIII Semana de Geoquímica*, Coimbra, pp. 180-82, viewed 02 nov 2020, .
- Simão, J., Leal, N. & Galhano, C. 2020, 'Chemical mobility of major elements during lixiviation experiments, in magmatic ornamental stones from Portugal', *Key Engineering Materials*, vol. 848, pp. 58-65, DOI:10.4028/www.scientific.net/ KEM.848.58
- Velbel, M.A. 1993, 'Constancy of silicate mineral weathering ratios between natural and experimental weathering: implications for hydrological control of differences in absolute rate', *Chemical Geology*, vol. 105, no. 1-3, pp. 89-99, DOI:10.1016/0009-2541(93)90120-8

- Velbel, M.A. 1999, 'Bond strength and the relative weathering rates of simple orthosilicates', *American Journal of Science*, vol. 299, no. 7-9, pp. 679-96, DOI:10.2475/ajs.299.7-9.679
- Viers, J., Dupré, B., Polvé, M., Schott, J., Dandurant, J.-L. & Braun, J.-J. 1997, 'Chemical weathering in the drainage basin of a tropical watershed (Nsimi-Zoette site, Cameroon): comparison between organic-poor and organic-rich waters', *Chemical Geology*, vol. 140, pp. 181-206.
- Shen, X., Arson, C., Ferrier, K.L., West, N. & Dai, S. 2019, 'Mineral weathering and bedrock weakening: modeling

microscale bedrock damage under biotite weathering', *Journal* of *Geophysical Research Earth Surface*, vol. 03, pp. 2623-46, DOI:10.1029/2019JF005068

- White, A.F. & Blum, A.E. 1995, 'Effects of climate on chemical weathering in watersheds', *Geochimica et Cosmochimica Acta*, vol. 59, no. 9, pp. 1729-47, DOI:10.1016/0016-7037(95)00078-E
- White, A.F. & Brantley, S.L. 2003, 'The effect of time on the weathering of silicate minerals: why do weathering rates differ in the laboratory and field?', *Chemical Geology*, vol. 202, no. 03-04, pp. 479-506, DOI:10.1016/j.chemgeo.2003.03.001

Author contributions

Ely Brasil de Arruda Luna Cavalcanti: writing; experimental analysis. José de Araújo Nogueira Neto: review and editing. Eldemar de Albuquerque Menor: review and editing. Adejardo Francisco da Silva Filho: review and editing. Marcelo Reis rodrigues da Silva: technical support. Lucas Fontenele Amorim: formatting.

Conflict of interest The authors declare no potential conflict of interest.

Data availability statement

All data included in this study are publicly available in the literature.

Funding information Not applicable.

Editor-in-chief Dr. Claudine Dereczynski

Associate Editor Dr. Márcio Fernandes Leão

How to cite:

Cavalcanti, E.B.A.L, Nogueira Neto, J.A., Menor, E.A., Silva Filho, A.F., Silva, M.R.R., Amorim, L.F. & . 2023, 'Simulated Acidic Weathering of "Black Granites": an Assessment using Regression Analysis', Anuário do Instituto de Geociências, 46:39560. https://doi.org/10.11137/1982-3908_2023_46_39560